

NEXT-TIME QUESTION

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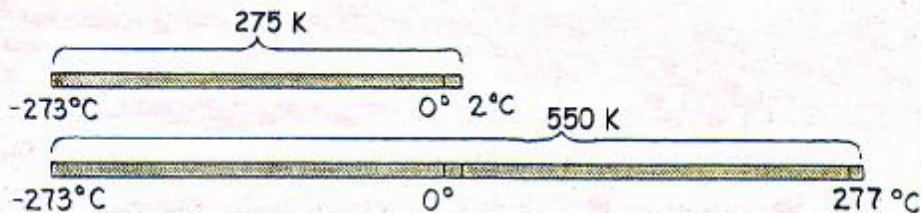
Answer: 277°C

Its temperature will be 277°C , and most certainly not 4°C !

At twice the internal energy, the gas will have twice the absolute temperature. Its initial absolute temperature is $273\text{ K} + 2\text{ K} = 275\text{ K}$. Twice this is 550 K . Expressed in Celsius, $550^{\circ}\text{C} - 273^{\circ}\text{C} = 277^{\circ}\text{C}$.



Consider a stick that is $273 + 2 = 275$ units long. This is like a thermometer that extends from absolute zero (-273°C) to 2°C . Can you see that a stick twice as long is $2(275)$ units long?



Or temperature-wise, 550 K ? Subtract the 273 part and you have 277 units - likewise for the twice-as-hot helium.